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Abstract

Iron based nanoparticles are one of the most applicable and studied nanostructures in various sciences and technologies. These nanoparticles are also introduced to the pharmaceutical and biomedical sciences due to their unique physicochemical properties such as super par magnetism, ease of synthesis, and biocompatibility. Several techniques are now developed and available for the preparation of iron based nanoparticles. Including chemical synthesis and biological synthesis which can be divided into microbial synthesis and plant mediated synthesis. Coprecipitation reaction is introduced as the main chemical method for the synthesis of magnetite nanoparticles are also reviewed as microbial and plant mediated synthesis. In this mini review we are going to have an over view upon the main approaches for the synthesis of iron nanoparticles.

Keywords: Biological synthesis, Biosynthesis, Chemical synthesis, Iron nanoparticles, Iron nanostructures.

1. Introduction

Iron nanoparticles are one of the most applicable nanostructures in pharmaceutical sciences and biomedicine. Theses nanoparticles are unique due to their physicochemical properties, biocompatibility, easy of synthesis and functionalization (1-7). Iron nanoparticles can be used in tissue engineering, magnetic resonance imaging (MRI), DNA detection, DNA and cell separation, intracellular labeling, magnetic transfections, antiproliferative hyperthermia therapy, and targeted drug delivery (8). Due to vast applications of iron nanoparticles in science and technology, various approaches have been developed for the synthesis of these nanoparticles. Chemical synthesis is one of the first approaches for the synthesis of iron nanoparticles. This production technic is the most defined and reproducible procedure with exact control over the synthesis process (9-12). But, this method employs harsh conditions such as protected atmosphere, high temperature and extreme pH (1, 2). Chemical synthesis protocols usually use organic solvents and toxic chemicals that may accumulate on the surface of the prepared nanoparticles and limit their medicinal and biomedical applications (1, 2).

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In recent years, biological approaches have been introduced as a novel and environmental friendly technique to address chemical synthesis limitations (13-22). Biological compounds from living organisms are able to produce biocompatible nanoparticles in mild synthesis conditions without using toxic chemicals (22). Also, use of biological compounds reduces the synthesis costs and also is economic for scaling up. To dates, microorganisms and plants have been used for biosynthesis of nanoparticles (13, 21, 22). Main microorganisms are bacteria and fungi, which may be pathogenic to human (23-26). So, nowadays, algae have attracted significant interest to produce nanoparticles (13). Bioactive compounds from plants are also useful for the synthesis of iron based nanoparticles in a sustainable manner (22). This mini review is dealing with most common chemical and biological procedures for the production of iron nanoparticles.

2. Chemical synthesis

Magnetite (Fe_3O_4) nanoparticles are the most applicable iron nanoparticle in pharmaceutical sciences and biomedicine. These nanoparticles are used in drug delivery, magnetic resonance imaging (MRI), and hyperthermia therapy (27-37). Magnetite nanoparticles are commonly synthesised by coprecipitation reaction of ferric ions (usually from ferric chloride) and ferrous ions (usually from ferrous chloride or ferrous sulphate) as in Eq. (1). The synthesis reaction should be done in a protected atmosphere and in the absence of oxygen to protect prepared nanoparticles from oxidation (3, 4, 38, 39). Magnetite nanoparticles can be oxidised into more oxidation states of iron such as maghemite (Fe_2O_3), which is not more useful for mentioned applications. Iron nanoparticles are usually coated by using biocompatible coatings such as silica, amino silanes, amino acids, lipoamino acids, carbohydrates, and polyethylene glycol (3, 6, 7, 11, 20, 40, 41). Biocompatible coatings increase the biocompatibility of iron nanoparticles and also increase the chemical and colloidal stability of the nanoparticles (34, 39, 42-48). Coating of nanoparticles can be done in two main approaches, separate reactions and one-put reaction. In the separate reactions, naked nanoparticles are

synthesised and then are coated in a separate reaction. This approach is time and labour consuming including various steps of synthesis, separation and washing (4). But, in a one-put reaction both synthesis of nanoparticles and coating can be done in just one reaction. In this approach, coprecipitation of ferrous and ferric ions is done in the presence of coating materials (3, 4, 6, 7, 9-12, 38-41, 49-54). It has been shown that presence of a coating compound in the synthesis reaction has immense impacts on the physicochemical properties of prepared particles. The major impact is on the particles size and particle size distribution. Presence of a coating material in the synthesis reaction provides more uniform nanoparticles with narrower particle size distribution (3).

$$\operatorname{Fe}^{2+} + 2\operatorname{Fe}^{3+} + 8\operatorname{OH}^{-} \rightarrow \operatorname{Fe}_{3}\operatorname{O}_{4} + 4\operatorname{H}_{2}\operatorname{O}$$
 (Eq.1)

3. Biological synthesis

Iron nanoparticles can also be synthesised in biosynthesis reactions by using various living organisms such as plants, algae, and microbial cells such as bacteria, fungi, and microalgae (20-23). Microbial cells can produce intra or extra cellular iron nanoparticles. Magneto tactic bacteria are one the most studied iron nanoparticles producing bacteria. These bacteria are capable to produce magnetite nanoparticles in their cytoplasm. Magnetite nanoparticles which are produced by magneto tactic bacteria are surrounded by a phospholipid bilayer membrane and known as magnetosome. An array of magnetosomes is arranged in the cytoplasm of the magneto tactic bacteria and makes a biologic compass for the cell (55-57). Bacterial cells are also able to produce iron nanoparticles extracellularly. They can accumulate iron nanostructures in their exopolysaccharide and produce a metal shell around the cell (23-26). It has been shown that iron nanoparticles which are synthesised extracellularly are ferrihydrite (a hydrous ferric oxyhydroxide) (23-26). Macro and micro algae are also able to produce iron nanoparticles. It is interesting that all iron nanoparticles which are reported to be synthesised by using seaweeds are magnetite nanoparticles (58-61).

Cell culture problems such as culture equipment and costs are major problem with

application of microbial cell for biosynthesis purposes. Previous investigations have been shown that plants are full of bioactive compounds such as antioxidant, phenolic compounds, carbohydrates and proteins which are able to interact with metallic ions and convert them to nanoparticles (22). A diverse kind of iron nanoparticles such as iron complexes, iron oxides, iron oxide hydroxides, and zero valent iron nanoparticles have been reported to be produced by plant mediated synthesis (22). Since now, various parts of plants such as leaf extract, fruit extract, seed exudate, seed powder extract, peel extract, bran extract, bark extract and plant oil have been used for synthesis of nanoparticles (62-85). A divergent variety of iron salts such as FeCl₂.6H₂O, FeCl₂.4H₂O, $FeSO_4$, and $Fe(NO_3)_3$ can be used as iron precursor (21, 22, 75, 86).

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4. Conclusion

Iron nanoparticles have retained most attention in various sciences and technologies due to their unique physicochemical and biological properties. Especially in pharmaceutical and biomedical sciences magnetite nanoparticles are one of the most studied and applied nanoparticles. These nanoparticles with special coatings are now commonly being synthesised by coprecipitation reaction. Also, these particles can be coated during synthesis reaction by one-put reaction techniques. Synthesis of nanoparticles is now developed toward green chemistry to reduce toxic wastes and non-sustainable protocols. Bioactive compound from biologic organisms such as bacteria, fungi, microalgae, algae, and plant are now being used instead of toxic chemicals to produce more safe and biocompatible nanoparticles.

Conflict of Interest

None declared.

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